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Soil Treatment Compositions and their Use

This invention relates to compositions for soil treatment which can provide both fertilisation and soil stabilisation benefits whilst being easy to handle using conventional equipment, and their use in irrigation methods.

It is well known in agriculture to apply various inorganic fertilisers to growing areas. The growing areas may be crop areas, which can be very large, or smaller growing areas such as those in greenhouses. The fertilisers can be supplied in various forms, in particular as solid compositions, or as suspensions or solutions of the fertiliser in a liquid. Fertiliser solutions are generally supplied by the manufacturer as an aqueous concentrate in large batches of size around 1 ton in weight. The solutions contain high concentrations, often 10 to 80 wt.% (dry solids), of dissolved inorganic fertiliser. It is usual to dose the concentrated fertiliser solution into irrigation water so that the solution is diluted and the fertiliser is applied to the soil during irrigation.

Various irrigation systems have been devised for convenient dosing of concentrated fertiliser solution. For instance, large scale irrigation is often carried out by spraying water onto a crop area. Spray pumps are well known which spray water from a spray manifold onto the crop area and which are designed so that concentrated fertiliser solution can be dosed into the pump, mixed with the water and thus diluted before the water is sprayed onto the crop area.

It is also known that some irrigation systems such as furrow irrigation and spray irrigation, as well as naturally occurring processes, can lead to erosion of the soil. It is known to use various polymers as soil stabilisers by applying them to the soil in aqueous solution. This stabilises the soil against erosion. Polymers of acrylamide and other ethylenically unsaturated monomers have been used as soil stabilisers. It has been generally accepted that the polymers which give optimum soil stabilisation have relatively low anionic content (eg 5 to 30 wt.%) and also have relatively high intrinsic viscosity, for instance above 15 dl/g. Such polymers tend to form viscous aqueous solutions unless used at low concentration. Normal industrial practice is to mix the polymer in powder or reverse phase emulsion form with water in special dissolution apparatus so as to form an aqueous solution of the polymer having a concentration of, for instance, 0.2 to 2 wt.%. This solution can then be

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dosed into the irrigation water.

US 3,798,838 describes a method of reducing the water permeability of soils by contacting them with an aqueous solution containing a water-soluble nutrient and a water-soluble partially hydrolysed polyacrylamide. The solution which is contacted with the soil may not contain more than 2 wt.% polymer or 5 wt.% nutrient. Preferred amounts are much lower, for instance not more than 0.3 wt.% polymer and not more than 0.05 wt.% nutrient. In the examples a dilute solution of this type is produced. The concentration of polyacrylamide is 70 ppm. The level of nutrient is 1.15 wt.%. The publication states that the polymer and the nutrients should be dissolved in the irrigation water. Thus the polymer and nutrient would be added separately directly to the irrigation water to give the dilute solution applied to the soil. There is no suggestion to provide a more concentrated solution containing both components. If polymer were to be included in a solution with the concentrated aqueous solution of fertiliser, the amount of polymer required to give an appropriate concentration on the soil (after dilution to the degree required for the fertiliser) would be high, ie at least 2 wt.%. The polyacrylamide used in the Examples has 20 to 30% of its amide groups hydrolysed, ie 20 to 30% anionic content. We have found that polymers of this type produce a highly viscous solution at such concentrations. As a result the solution cannot be processed in the same apparatus as standard fertiliser solution. Therefore the system of US 3,798,838 is limited to separate addition of the polymer and the fertiliser to the irrigation water. This necessitates, in commercial practice, the provision of two separate addition points and make-up systems.

Other publications describe various compositions which contain both a conventional low anionic content soil conditioner and a fertiliser, such as EP-A-586,911 and JP-A-74/144087. JP-A-07/322753 and JP-A-07/099833 each describe soil conditioning compositions containing acrylamide polymers. Aqueous compositions containing polymer and fertiliser are not disclosed. JP-A-73/037738 also discloses a composition suitable for spraying comprising fertiliser, polyacrylamide and seeds, together with a collection of other materials such as cellulose and lignin sulphonic acid. JP-A-51/124578 also discloses a liquid fertiliser composition containing phosphorus fertiliser and a copolymer of acrylamide and unspecified amounts of potassium acrylate.

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It would be desirable to be able to provide a combined composition which gives both fertilisation and soil stabilisation benefits, so as to avoid the need for separate addition points, but to be able to use dosing equipment currently in place for irrigation systems, in particular equipment in place for processing fertiliser solution. Such systems include pivots, wheel-lines and drip systems.

According to the invention we provide an aqueous soil treatment composition comprising water and, in solution,

(a) an ionic (eg inorganic or urea-containing) water-soluble fertiliser at a concentration of at least 10 wt.%, and

(b) a water-soluble anionic polymer which has intrinsic viscosity at least 6 dl/g and is formed from water-soluble monomer or monomer blend of which at least 40 wt.% is anionic monomer.

Thus we provide a single pack product which gives both fertilisation and soil stabilisation benefits. The product is thus a growth-promoting composition. It can be produced by adding polymer to a standard concentrated solution of fertiliser, at a concentration sufficiently high to provide appropriate polymer concentration on the soil after dilution of the concentrated solution in the standard manner for fertiliser solutions. We find that the defined anionic polymers having relatively high anionic monomer content give rise to fewer of the viscosity problems which we observe with the standard, non-ionic or low anionic content, soil stabilisation polymers. In fact, in the solution of fertiliser they have relatively low viscosities which can even be below 1,000 cps, in particular below 500 cps. Thus the viscosity of the fertiliser solution is not increased to an inconvenient degree and the aqueous composition of the invention can be processed using the equipment which is in place for processing of solution of fertiliser alone. A further advantage of the defined polymers is that their low viscosity in the composition means that they can be added without difficulty to the fertiliser solution at the fertiliser production plant. The prevention of excessive viscosity allows the compositions to be processed at the fertiliser production plant using conventional equipment.

We believe that the interaction between the fertiliser which is an ionic salt, and the defined anionic polymer is important in the invention. We believe that the polymer dissolves in the aqueous composition but in the presence of the fertiliser salt

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the polymer molecules are not fully extended, thus reducing the viscosity of the composition. However, as the composition is diluted during an irrigation process by mixing with the irrigation water, the concentration of fertiliser is reduced and the polymer molecules are able to extend fully and act as soil stabilisers. Similarly, if the composition is placed directly onto an area of soil, the fertiliser gradually dissolves into the soil, thus reducing its concentration around the polymer molecules, which can again extend and act as soil stabilisers.

The invention provides an aqueous composition in which the defined polymer (b) is dissolved. That is, the polymer is taken into the solution such that substantially no visible solid material remains. For some combinations of polymer and fertiliser, some solid lumps or other material visible to the naked eye may remain. These may be filtered out before use. It is not necessary that the polymer molecules be fully extended and solvated within the solution. In fact, we believe that it is useful in the invention that the polymer molecules, although dissolved, are not fully extended in the solution. They thus give rise to lower viscosity than they would if fully extended.

The polymer must be water-soluble and in particular is soluble in an aqueous solution of fertiliser having the same concentration of fertiliser as the final aqueous composition which is desired to be produced. Generally the polymer is substantially linear and is not cross-linked.

The polymer has intrinsic viscosity at least 6 dl/g. In this specification intrinsic viscosity is measured by suspended level viscometer at 20°C in 1 M sodium chloride buffered to pH 7. That is, it is of sufficiently high molecular weight to give a soil stabilisation effect and is not a low molecular weight material which would act as a dispersant.

Preferably IV is at least 8 dl/g, more preferably at least 9 dl/g. It may be up to for instance 30 dl/g but generally we find that the optimum combination of low viscosity of the composition and soil stabilisation performance is given by polymers having IV not more than about 20 or 18 dl/g. Preferably IV is not more than 16, more preferably not more than 15 dl/g. Particularly preferred IV ranges are from 9 to 13 dl/g, especially 10 to 12 dl/g.

The polymer is formed from water-soluble monomer or monomer blend, usually water-soluble ethylenically unsaturated monomer. The anionic content, ie the

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proportion of anionic monomer in the monomer blend used to form the polymer, is at least 40 wt.%, preferably at least 45 or at least 50 wt.%. In particular it is at least 55 wt.%. It can be up to 100 wt.%, but is preferably not more than 90 wt.%, more preferably not more than 80 wt.%. Particularly preferred polymers have anionic content in the range 65 to 85 wt.%.

The monomer blend used to form the polymer comprises at least 40 wt.% anionic monomer. This may be any suitable anionic ethylenically unsaturated monomer. It can be a sulphonic monomer, often as sodium or other alkali metal salt, for instance acrylamido propane sulphonic acid (AMPS). It is generally preferred that the anionic monomer is an ethylenically unsaturated carboxylic monomer, in particular acrylic or methacrylic monomer. Salts of acrylic acid are preferred, for instance ammonium or alkali metal, in particular sodium, salts.

The polymer may contain small amounts of cationic monomer, for instance up to 20 wt.% or 10 wt.% but usually the content of cationic monomer is substantially zero.

Generally the anionic monomer is copolymerised with non-ionic monomer, usually ethylenically unsaturated water-soluble non-ionic monomer such as acrylamide or methacrylamide, preferably acrylamide.

Particularly preferred polymers are copolymers of acrylamide with sodium acrylate.

The polymer is included in the composition in an amount such that when the composition is diluted for irrigation, to the degree of dilution conventionally used for fertiliser solutions, it provides adequate polymer concentration on the soil to give acceptable soil stabilisation performance. Polymer concentration in the aqueous soil treatment composition of the invention is generally from 0.2 to 10 wt.%, in particular at least 0.5 or at least 1 wt.%. Preferably it is at least 1.5 wt.%. Often it is not more than 7 wt.%, and particularly preferred compositions contain from 2 to 5 wt.% polymer.

The composition also contains an ionic fertiliser in solution. Thus the fertiliser must be water-soluble. Fertilisers for use in the invention can also be described as inorganic or urea-containing fertilisers. The fertiliser is present in a concentration at least 10 wt.% and is usually as high as is convenient, for instance at least 20 wt.%.

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Preferably it is at least 30 wt.%. It can be as high as 70 or 80 wt.% but is usually not more than 60 wt.%. Preferred concentration ranges are 30 or 40 to 55 wt.%, for instance around 50 wt.%.

Urea and any of the known inorganic fertiliser materials can be used, which provide nitrogen, phosphorus and/or potassium, either alone or in a mixture. These contain ionic salts and include ammonium nitrate, ammonium sulphate, mono ammonium phosphate, diammonium phosphate, mono potassium phosphate, dipotassium phosphate, polyphosphate salts, potassium chloride, potassium sulphate and calcium nitrate.

Examples include urea/ammonium nitrate (32-0-0), potassium chloride (0-0-10), ammonium sulphate (8-0-0-9S), a blend of calcium and ammonium nitrate, and blended fertilisers with the following analyses: 18-0-8, 10-0-10, 3-18-18, 0-0-25-17S, 10-10-10 and 14-2-10-2+2.5% organic matter (OM).

The invention is particularly useful when it is known to be desired to use a particular fertiliser on the relevant soil area. The invention allows the operator to select a polymer which gives, in combination with that chosen fertiliser, an appropriate balance of viscosity of the final composition and soil stabilisation performance. We find that any given polymer will not necessarily give optimum results for all fertilisers, but generally we find that polymers of the types defined in the invention are capable of giving an acceptable range of results in combination with a range of fertilisers. The particular intrinsic viscosity and anionic content within the claimed ranges which are optimum for combination with any given fertiliser, as well as the concentration of polymer and the concentration of fertiliser, can be established by experimentation.

In particular we find that for use with UAN (32-0-0) solution polymers of anionic content 60 to 70 wt.% and IV 9 to 11 dL/g, for instance anionic content 65 wt.% and IV 10 dL/g, give low viscosity and good soil stabilisation performance. Polymers of higher anionic content, for instance from 90 to 100 wt.%, and slightly higher intrinsic viscosity, for instance 12 to 14 dL/g, give higher viscosity with the same fertiliser solution although the compositions can still be handled, and show particularly good stabilisation performance.

Polymers of anionic content 60 to 70 wt.% and IV 9 to 11 dL/g give low

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The invention will now be illustrated with reference to the following examples.

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Examples

In the following examples various fertilisers were used, as follows:

Urea/Ammonium Nitrate (32-0-0)

Muriate of Potash (0-0-10)

Sulfate of Ammonia (8-0-0-9S)

Calcium Ammonium Nitrate

Blended fertilisers with analysis:

18-0-8

10-0-10

3-18-18

0-0-25-17S

10-10-10

14-2-10-2+2.5% OM

Example 1

Several polymers, of medium to high anionic charge and in dry powder form, were evaluated initially for their solubility and viscosity in concentrated solutions of fertiliser. This was assessed by adding the polymer powder to the fertiliser solution and tumbling for 4 to 5 hours at slow medium speed. The polymer powder was added in such an amount as to give a concentration in the composition of about 2 wt.%.

The viscosities of the resultant products were measured using a Brookfield LVT Viscometer, spindle 4 at 30 rpm.

If the resultant product was not highly viscous, ie was pourable, the flocculation and stabilisation tests were performed. Both of these tests give an indication of the soil stabilisation efficacy of the relevant polymer in combination with the relevant fertiliser. These tests were evaluated using 34 ppm hardness water, to simulate the irrigation water commonly used in the USA (a region in which large-scale irrigation is widely practised).

The polymers used were as follows:

Polymer A: Copolymer of 14.6 wt.% sodium acrylate and 85.4 wt.% acrylamide, IV 16 dl/g.

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Polymer B: Copolymer of 65 wt.% sodium acrylate and 35 wt.% acrylamide, IV 10 dl/g.

Polymer C: Copolymer of 95.5 wt.% sodium acrylate and 4.5 wt.% acrylamide, IV 13 dl/g.

Polymer D: Copolymer of 40 wt.% sodium AMPS and 60 wt.% acrylamide, IV 18 dl/g.

Polymer E: Copolymer of 20 wt.% sodium AMPS and 80 wt.% sodium acrylate, of low IV.

The fertiliser solutions were all used as supplied by the manufacturer, with varying solids content of from 10 to 80 wt.% solids.

Flocculation

The 34 ppm hardness water was added to a 1,000 ml stoppered measuring cylinder (~950 g). The solution of polymer and fertiliser was then added at 15 ppm before the remainder of the water was added, to total 1,000 g. The cylinder was inverted several times to allow the product to disperse. Soil fines (< 500 mm size particles of a clay loam soil), 20 g, were added to the cylinder and the cylinder shaken vigorously for 20 seconds, before leaving to stand for a further 2 minutes, to allow flocculation to occur. An aliquot was then extracted from a predetermined depth and the turbidity measured, to assess the effectiveness of the product. The clearer the water, the more effective the product.

Stabilisation

The test above was repeated, using standardised (sieved and dried) soil crumbs and not fines. In this test, the cylinder was not shaken vigorously, but was inverted 10 times in 20 seconds, before leaving to stand for 2 minutes. The turbidity was then measured.

A control was also performed for the flocculation and stabilisation tests, where the water alone was used, without any polymer present.

The results in Table 1 below show the effect of the polymers on flocculation/soil stabilisation. The stabilisation test was only performed on products which were pourable and which produced good flocculation results.

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Table 1

Fertiliser	2% Polymer	Viscosity (cPs)	Turbidity NTU (Flocculation)	Turbidity NTU (Stabilisation)
Control	---	---	758.7	135
UAN 32	A'	11,030	171.3	---
UAN 32	B	1,510	216.3	38.3
UAN 32	C	4,140	177.3	52
UAN 32	D	6,780	528	---
8-0-0-9S	B	640	285.3	---
8-0-0-9S	C	3,790	76.7	18
8-0-0-9S	E	90	798	---

It can be seen from this example that Polymer B, which has anionic content of 65 wt.% and IV 10 dl/g, is particularly suitable for providing low viscosity, and thus pumpable and sprayable, compositions. In particular it gave adequately low viscosity with two different fertilisers, UAN 32 and 8-0-0-9S. In this particular test, flocculation in the 8-0-0-9S solution was not optimum.

It can also be seen that Polymer C, which has a higher anionic content of 95.5 wt.%, gives higher viscosities than Polymer B, but lower than those of Polymers A and D in UAN 32 solution. In addition, it gives good flocculation and stabilisation performance.

Polymers A and D gives rather high viscosity values and Polymer E does not give good flocculation.

Example 2

In this Experiment the effects of the IV and anionic content of various polymers in various fertiliser solutions are assessed.

FED Experiment

A factorial experiment design, using solid grade acrylamide/sodium acrylate polymers (added at 2% w/w to the concentrated fertiliser solution), covering a range of IV's from 10 to 24 and anionic contents of 0 to 100%, was evaluated. In each case, the viscosity and flocculation of the resultant solutions were measured as in Example 1 above. This was performed with a potassium chloride solution (10% K⁺), which was similar to the 0-0-10(KCl) fertiliser solution. A control was performed, using 15 ppm of the KCl solution, without added polymer.

The factorial experimental design results are in Table 2 below. Some products are listed twice. This is because two different batches were assessed. It will be seen that polymers which can give rather high viscosities can also give variable results from batch to batch.

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Table 2

Polymer	Anionic (%)	IV	Viscosity (cPs)	Flocculation (NTU)
Control	---	---	---	---
F	28	12	5,050	645
G	14.25	18	12,500	327
B	65	10	350	126
B	65	10	375	240
H	47	11	1,250	148
H	47	11	3,575	281
C	95.5	13	4,900	248
I	75	20	7,000	120
A	14.6	16	8,625	180
J	48	24	11,375	232
Q	0	16	8,750	407
K	14.9	24	V.Viscous	39
L	42	16	5,975	---
L	42	16	6,150	284
H	47	16	5,600	324
H	47	16	9,000	216
N	84	19	6,500	267
P	100	13	6,750	150
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It can be seen from these results that Polymer B again gives a low viscosity and good flocculation performance this fertiliser solution.

Polymers F, G, A, Q and K all have anionic content below 40 wt.% and give rather high viscosity values, in particular Polymers G, A, Q and K. From the point of view of achieving low viscosity in combination with adequate flocculation performance, it appears that a value of IV of from about 9 or 10 to about 13 or 14 tends to be optimum in combination with anionic content of at least 40 wt.%. It also appears that anionic content of from about 55 to 80 wt.% may be optimum.

These tests were carried out with a fertiliser solution similar to 0-0-10 (KCl) solution. Certain of the polymers may give better performance with other fertiliser types.

Example 3

In this Example flocculation tests were also performed on the optimum polymers determined in Example 1 for each fertiliser, using a different soil type. These were assessed using soil of a loamy sand nature.

The optimum products to use in UAN (32-0-0) and 8-0-0-9S fertilisers were found in Example 1 to be Polymer B and Polymer C respectively. These were evaluated with a second soil type, a loamy sand. The 8-0-0-9S was also assessed with Polymer B, as this produced a much less viscous solution (640 cPs), than Polymer C (3790 cPs). Results were as follows.

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Table 3

Fertiliser & Polymer	Flocculation (NTU)	Control (NTU)
KCl + B	201 (194)*	519
UAN + B	267 (216)*	574
8-0-0-9S + C	78 (77)*	507
8-0-0-9S + C	335 (285)*	507

* (The figures in brackets are the original turbidity results for flocculation, when performed on the first soil type in Example 1, a clay loam).

It can be seen that although flocculation was in general not as efficient as on the previously used soil, it was nevertheless effective.

Example 4

The solubilities of Polymer B and Polymer C (the optimum polymers determined in Example 1) were assessed at 2% w/w in various fertiliser solutions. The viscosities of the polymers in fertiliser solutions were measured, if the polymers were soluble, and so was the flocculation efficiency. Results are below in Table 4.

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Table 4

Fertiliser	2% Polymer	Viscosity (cPs)	Flocculation
18-0-8	B	1,000	155
18-0-8	C	3,600*	---
10-0-10	B	900	151
10-0-10	C	3,950	124
10-10-10	B	950	306
10-10-10	C	3,850	297
0-0-25-17S	B	*	
0-0-25-17S	C	10,000*	---
14-2-10-2+2.5% OM	B [Mag 525]	1,500	205
14-2-10-2+2.5% OM	C [Alclar W5]	7,500	---

* - large quantity of undissolved polymer after tumbling overnight.

Both products were insoluble in the 3-18-18 fertiliser solution.

These results again demonstrate the general usefulness of Polymer B, which gave low viscosity solutions and good flocculation for a range of fertilisers. Polymer C could also give reasonable viscosity with certain fertilisers such as 18-0-8, 10-0-10 and 10-10-10.

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